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The video below refers to Dynamic Equilibrium (under the section Equilibrium Systems) 7.1 p. 420 in the Nelson Textbook The video below is related to 7.2 Equilibrium Law and the Equilibrium Constant (p.429 in the Textbook)

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Tuesday August 12, 2014 . Review Questions, p. 39, #1-6. Define Carbonyl group, and aldehyde and ketone, Name IUPAC (Remember International Union of Pure and Applied Chemistry) Wednesday August 13, 2014 . Assessment on Naming of Alkanes, Alkenes, and Alkynes. Thursday August 14, 2014 . 1.6 Carboxylic Acids, Esters, and Fats . Define each one ...

### Pre University Courses

= 12.00 mmol - 5.40 mmol Unreacted ethanoic acid = 6.60 mmol Since 18.00 mL of NaOH(aq) was added to 20.00 mL of ethanoic acid solution, the total volume is now 38.00 mL. Solution: The concentration of unreacted ethanoic acid, [HC 2H 3O 2(aq)], can be determined: 23 2 23 2 HC H O (aq) 23 2 HC H O (aq) 23 2 [HC H O (aq)] (6.60 mmol) = 38.00 mL

### Section 8.7: Acid-Base Titration Tutorial 1 Practice, page 547

2 Chemistry, First Canadian Edition 1.17 An organic compound consists of molecules made from carbon, hydrogen and oxygen atoms. There can be other atoms – for example, nitrogen, sulfur. 1.19 1. Acetylsalicylic acid, the common mild pain reliever, is a derivative of the natural product, salicin, obtained from willow bark. 2.

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Chemistry 12 - Chapter 3 Quiz. True/False. Indicate whether the sentence or statement is true or false. 1. The region in space where an electron is most likely to be found is called an energy level. 2. Electron configurations are often condensed by writing them using the previous noble gas core as a starting point. ...

### Chemistry 12 - Chapter 3 Quiz - Nelson

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